Polish J. Chem., 76, 623–625 (2002)

Interaction of the Components in Ti-{Si, Ge}-Bi Systems at 670 K

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(Received July 27th, 2001; revised manuscript November 12th, 2001)

The interaction of titanium with silicon or germanium and bismuth over the whole concentration range has not yet been studied. Therefore, the binary systems of the Ti–{Si, Ge}–Bi ternary systems have been widely studied, including the phase diagrams over the whole concentration range. Before investigation of ternary equilibrium diagrams, we confirmed the existence of the binary compounds. Two intermetallic phases exist in Ti–Bi system at 670 K [1]. Ti₂Bi (Ti₂Bi structure type) [2,3] intermetallic compound has been found at 66.7 at. % of Ti. An intermetallic phase, which has been found at ~75 at. % of Ti is in equilibrium with this compound and pure titanium. Auer-Welsbach *et al.* [3] and Nowotny *et al.* [4] were the first to report the existence of this compound of the composition Ti₄Bi, based on a structural similarity to Ti₄Pb (hexagonal Ni₃Sn type of structure). According to the metallographic and X-ray investigations [5], however, this phase has the stoichiometry Ti₃Bi with a tetragonal structure. We could not determine the crystal structure of the phase, which we found, because we obtained a poor quality diffractogram. Both of these compounds (Ti₂Bi and Ti₃Bi) are characterized by constant compositions.

Five binary compounds exist in Ti–Si system at 670 K [1,6,7]. Ti₃Si (Ti₃P structure type), Ti₅Si₄ (Zr₅Si₄ type), TiSi (FeB type) and TiSi₂ (TiSi₂ type) compounds have stable compositions. Ti₅Si₃ (Mn₅Si₃ type) possesses the homogeneity range of 36.5-38.5 at. % Si.

Three binary compounds exist in Ti–Ge system at 670 K [1,6,8]. These are Ti_5Ge_3 (Mn_5Si_3 structure type), Ti_6Ge_5 (Ti_6Ge_5 type), $TiGe_2$ ($TiSi_2$ type). We could not find Ti_3Ge (Ni_3P type) and TiGe (anti-TiSi type). All these compounds are characterized by constant compositions. Crystallographic parameters for the binary phases, existing in these systems, are given in Table 1. No binary compounds exist in Si–Bi and Ge–Bi systems. The Si–Bi system is characterized by a monotectic transformation at 1673 K and an eutectic transformation close to the melting point of bismuth. The Ge–Bi system is characterized by an eutectic transformation very close to the melting point of bismuth [1].

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Table 1. Crystallographic parameters for the binary phases in Ti-{Si, Ge}-Bi systems.

In this communication we present the results for the isothermal section of Ti-Si-Bi system and Ti-Ge-Bi system at 670 K. The samples were prepared by reaction of the elements. High purity metals (titanium – 99.99; silicon – 99.999; germanium – 99.999; bismuth – 99.99) were used. Alloys have been prepared by arc melting under a protective argon atmosphere (with Ti as a getter), using a nonconsumable tungsten electrode. During arc melting the weight losses were less than 1.5%. The alloys were annealed in quartz ampoules under vacuum at 670 K for 360 h.

Isothermal section of the phase diagrams of Ti–Si–Bi and Ti–Ge–Bi systems have been constructed by X-ray phase analysis. Powder patterns for the phase analysis



Figure 1. Isothermal section of Ti–Si–Bi system at 670 K.



Figure 2. Isothermal section of Ti-Ge-Bi system at 670 K.

were obtained on powder diffractometers DRON-2.0 (FeK_{α}-radiation) and DRON-3M (CuK_{α}-radiation, 0.05° step of scanning, 10 sec./one point speed of scanning). Indexing of the diffractograms has been made by the program TREOR-90. The cell constants have been refined using the program LATCON.

By comparing and analysing the X-ray diffraction patterns of 72 samples and identifying the phases in each sample, the isothermal section at 670 K of the phase diagrams of the $Ti-{Si, Ge}-Bi$ ternary systems has been determined. For the Ti-Si-Bi ternary system at 670 K it is shown in Fig. 1 and that of Ti-Ge-Bi - in Fig. 2. No ternary compounds and solid solutions on the bases of binary compounds have been found.

REFERENCES

- 1. Massalski T.B., Binary Alloy Phase Diagrams, Am. Soc. for Met. Metal Park, Vol. 1-3 (1986).
- 2. Parthe E., Gelato L., Chabot B. *et al.* TYPIX. Standartized Data and Crystal Chemical Characterization of Inorganic Structure Types//Springer-Verlag, Berlin, 1993. Vol. 1–4. 1596 p.
- 3. Auer-Welsbach H., Nowotny H. and Kohl A., Monatsh. Chem., 89, 154 (1958).
- 4. Nowotny H. and Pesl J., Monatsh. Chem., 82, 344 (1951).
- 5. Obinata I., Takeuchi Y. and Saikawa S., Trans. ASM, 52, 1059 (1960).
- 6. Villars P. and Calvert L.D., Pearson's Handbook of Crystallographic Data for Intermetallic Phases, ASM-International (1997).
- 7. Seifert H.J., Lukas H.L. and Petzow G., Thermodynamic Optimization of the Ti–Si System, Z. Metallkd. 1996. Vol. 87, P. 2–13.
- 8. Jain T.A. and Kao C.R., Binary Compounds in the Ge–Ti System, J. Alloys Comp., 282, p. L9–L12 (1999).